## Work through the ChemCAL modules "Alkenes, Benzene and Alkynes" and "Elimination and Electrophilic Addition Reactions".

1. Give the stick structures of the following compounds.
(a) 2,3,5-trimethyl-4-bromooctane
(b) cis-1,3-dimethylcyclobutane
(c) 2-methyl-2-pentene
2. Name the following compounds.



3. Draw the structures of the following compounds.
(a) 3-chloro-2,3-dimethyl-1-butene
(b) 3-ethyl-3-hexene
(c) (Z)-2-butene
(d) (E)-2-butene
(e) (Z)-2-methyl-3-hexene
4. All atoms, bonds and lone pairs are shown in the structures below. Use your knowledge of valency and arrow notation to add formal charges, $\oplus$ and $\ominus$, on the structures where it is appropriate. Add partial charges, $\boldsymbol{\delta} \oplus$ and $\delta \ominus$, to the neutral reagents.

5. Classify the two starting materials (below) as electrophile, nucleophile or neither. What kind of reaction is this? Draw in appropriate partial charges $\delta \oplus$ and $\delta \ominus$ and curly arrows showing the mechanism of the reaction.

